

Bulkwholesale Australia Pty Ltd Chemwatch Hazard Alert Code: 3 Chemwatch: 5498-26 Issue Date: 12/10/2021 Version No: 2.1 Print Date: 10/06/2022 Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements S.GHS.AUS.EN

SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier	
Product name	Bulk Blendz Tyre Gel
Chemical Name	Not Applicable
Synonyms	Not Available
Chemical formula	Not Applicable
Other means of identification	Not Available

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Use according to manufacturer's directions.
	bee decording to manafacturer o anocatorio.

Details of the supplier of the safety data sheet

Registered company name	Bulkwholesale Australia Pty Ltd
Address	2/7 Commercial Court, Tullamarine VIC 3043 Australia
Telephone	1300 096 435
Fax	
Website	https://www.bulkwholesale.com.au
Email	orders@bulkwholesale.com.au

Emergency telephone number

Association / Organisation	N.V.Chemicals(Aust) P/L	CHEMWATCH EMERGENCY RESPONSE
Emergency telephone numbers	0411 387 097	+61 1800 951 288
Other emergency telephone numbers	Not Available	+61 3 9573 3188

Once connected and if the message is not in your prefered language then please dial 01

SECTION 2 Hazards identification

Classification of the substance or mixture

Poisons Schedule	Not Applicable
Classification ^[1]	Skin Corrosion/Irritation Category 2, Serious Eye Damage/Eye Irritation Category 1, Hazardous to the Aquatic Environment Long-Term Hazard Category 3
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI

Label elements	
Hazard pictogram(s)	
Signal word	Danger

H315	Causes skin irritation.
H318	Causes serious eye damage.
H412	Harmful to aquatic life with long lasting effects.

Precautionary statement(s) Prevention

P280	Wear protective gloves, protective clothing, eye protection and face protection.
P273	Avoid release to the environment.
P264	Wash all exposed external body areas thoroughly after handling.

Precautionary statement(s) Response

P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P310	Immediately call a POISON CENTER/doctor/physician/first aider.
P302+P352	IF ON SKIN: Wash with plenty of water.
P332+P313	If skin irritation occurs: Get medical advice/attention.

Precautionary statement(s) Storage

Not Applicable

Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

Not Applicable

SECTION 3 Composition / information on ingredients

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
63148-62-9	10-<20	polydimethylsiloxane
6540-99-4	10-<20	lauryl alcohol, ethoxylated
7732-18-5	balance	water
Legend:	 Classified by Chernwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L * EU IOELVs available 	

SECTION 4 First aid measures

Description of first aid measures

Eye Contact	 If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 If skin contact occurs: Immediately remove all contaminated clothing, including footwear. Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation.
Inhalation	 If fumes, aerosols or combustion products are inhaled remove from contaminated area. Other measures are usually unnecessary.
Ingestion	 If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Seek medical advice.

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

SECTION 5 Firefighting measures

Extinguishing media

- ▶ There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area

Special hazards arising from the substrate or mixture

Fire Incompatibility

ility + Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

Advice for firefighters	
Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves in the event of a fire. Prevent, by any means available, spillage from entering drains or water courses. Use fire fighting procedures suitable for surrounding area.
Fire/Explosion Hazard	 High temperature decomposition products include silicon dioxide, small amounts of formaldehyde, formic acid, acetic acid and traces of silicon polymers. These gases may ignite and, depending on circumstances, may cause the resin/polymer to ignite. An outer skin of silica may also form. Extinguishing of fire, beneath the skin, may be difficult. carbon dioxide (CO2) silicon dioxide (SiO2) other pyrolysis products typical of burning organic material. May emit corrosive fumes. CARE: Water in contact with hot liquid may cause foaming and a steam explosion with wide scattering of hot oil and possible severe burns. Foaming may cause overflow of containers and may result in possible fire.
HAZCHEM	Not Applicable

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Environmental hazard - contain spillage. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb spill with sand, earth, inert material or vermiculite.
Major Spills	 Environmental hazard - contain spillage. Silicone fluids, even in small quantities, may present a slip hazard. It may be necessary to rope off area and place warning signs around perimeter. Clean up area from spill, with suitable absorbant, as soon as practically possible. Final cleaning may require use of steam, solvents or detergents. Moderate hazard. Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

Precautions for safe handling	
Safe handling	 DO NOT allow clothing wet with material to stay in contact with skin Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. Avoid contact with moisture.
Other information	 Store in original containers. Keep containers securely sealed. No smoking, naked lights or ignition sources. Store in a cool, dry, well-ventilated area.

Conditions for safe storage, including any incompatibilities

Suitable container	20Lt/5Lt. ► Polyethylene or polypropylene container. ► Packing as recommended by manufacturer. ► Check all containers are clearly labelled and free from leaks.
Storage incompatibility	 Avoid reaction with oxidising agents, bases and strong reducing agents. Avoid strong acids, acid chlorides, acid anhydrides and chloroformates.

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (DEL)		
INGREDIENT DATA			
Not Available			
Emergency Limits			
Ingredient	TEEL-1	TEEL-2	TEEL-3
polydimethylsiloxane	65 mg/m3	720 mg/m3	4,300 mg/m3

Ingredient	TEEL-1	TEEL-2	TEEL-3		
lauryl alcohol, ethoxylated	2.9 mg/m3	31 mg/m3	200 mg/m3		
Ingredient	Original IDLH		Revised IDLH		
polydimethylsiloxane	Not Available		Not Available		
lauryl alcohol, ethoxylated	Not Available	Not Available Not Available			
water	Not Available		Not Available		
Occupational Exposure Banding	1				
Ingredient	Occupational Exposure Ba	and Rating	Occupational Exposure Band Limit		
lauryl alcohol, ethoxylated	E		≤ 0.1 ppm	≤ 0.1 ppm	
Notes:	Occupational exposure band adverse health outcomes as range of exposure concentra	ling is a process of assigning chemical sociated with exposure. The output of t ations that are expected to protect work	s into specific categories or bands based on a chemical's potency an this process is an occupational exposure band (OEB), which correspo rer health.	d the onds to a	
Exposure controls					
Appropriate engineering controls	Engineering controls are use be highly effective in protect The basic types of engineeri Process controls which invol Enclosure and/or isolation of "adds" and "removes" air in	ed to remove a hazard or place a barrie ing workers and will typically be indepen ng controls are: Ive changing the way a job activity or pr f emission source which keeps a select the work environment.	er between the worker and the hazard. Well-designed engineering cor ndent of worker interactions to provide this high level of protection. rocess is done to reduce the risk. ed hazard "physically" away from the worker and ventilation that strat	tegically	
Personal protection					
Eye and face protection	 Safety glasses with side shields. Chemical goggles. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. 				
Skin protection	See Hand protection below				
Hands/feet protection	 Wear chemical protectiv Wear safety footwear or NOTE: The material may produ equipment, to avoid all p Contaminated leather its The selection of suitable glo manufacturer. Where the ch and has therefore to be chee The exact break through tim making a final choice. Personal hygiene is a key el 	e gloves, e.g. PVC. safety gumboots, e.g. Rubber ce skin sensitisation in predisposed ind possible skin contact. ems, such as shoes, belts and watch-ba ves does not only depend on the mater emical is a preparation of several subst sked prior to the application. e for substances has to be obtained fro ement of effective hand care.	lividuals. Care must be taken, when removing gloves and other prote ands should be removed and destroyed. rial, but also on further marks of quality which vary from manufacturer tances, the resistance of the glove material can not be calculated in a om the manufacturer of the protective gloves and has to be observed	ctive ⁻ to idvance when	
Body protection	See Other protection below				
Other protection	 Overalls. P.V.C apron. Barrier cream. Skin cleansing cream. 				

Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection:

NV Tyre Gel

Material	CPI
BUTYL	A
NEOPRENE	А
VITON	А
NATURAL RUBBER	С
PVA	С

* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

Respiratory protection

Type A-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Selection of the Class and Type of respirator will depend upon the level of breathing zone contaminant and the chemical nature of the contaminant. Protection Factors (defined as the ratio of contaminant outside and inside the mask) may also be important.

Required minimum protection factor	Maximum gas/vapour concentration present in air p.p.m. (by volume)	Half-face Respirator	Full-Face Respirator
up to 10	1000	A-AUS / Class1 P2	-
up to 50	1000	-	A-AUS / Class 1 P2
up to 50	5000	Airline *	-
up to 100	5000	-	A-2 P2
up to 100	10000	-	A-3 P2
100+			Airline**

* - Continuous Flow ** - Continuous-flow or positive pressure demand A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

- Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance	Blue liquid.		
Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Applicable
pH (as supplied)	Not Available	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	Not Applicable	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Not Available	pH as a solution (Not Available%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 Stability and reactivity

Reactivity	See section 7
Chemical stability	 Silicone fluids are stable under normal storage conditions. Hazardous polymerisation will not occur. At temperatures > 150 C, silicones can slowly react with the oxygen in air. When heated > 300 C, silicones can slowly depolymerise to volatile siloxanes whether or not air is present. Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 Toxicological information

Information on toxicological effects

Inhaled	The material is not thought to produce either adverse health effects or irritation of the respiratory tract following inhalation (as classified by EC Directives using animal models). Nevertheless, adverse systemic effects have been produced following exposure of animals by at least one other route and good hygiene practice requires that exposure be kept to a minimum and that suitable control measures be used in an occupational setting. Vapours of silicones are generally fairly well tolerated, however very high concentrations can cause death within minutes due to respiratory failure. At high temperatures, the fumes and oxidation products can be irritating and toxic and can cause depression leading to death in very high doses.
Ingestion	Accidental ingestion of the material may be damaging to the health of the individual. Silicone fluids do not have a high acute toxicity. They may have a laxative effect and produce central nervous system depression.
Skin Contact	This material can cause inflammation of the skin on contact in some persons. Low molecular weight silicone fluids may exhibit solvent action and may produce skin irritation. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

Page 6 of 9

Bulk Blendz Tyre Gel

Lye	Eye exposure to silicone fluids causes temporary irritatio	Eye exposure to silicone fluids causes temporary irritation of the conjunctiva. Injection into the specific structures of the eye, however, causes corneal scarring, permanent eye damage, allergic reactions and cataract, and may lead to blindness.		
Chronic	Corneal scarring, permanent eye damage, allergic reacti Prolonged or repeated skin contact may cause drying wi Substance accumulation, in the human body, may occur There is limited evidence that, skin contact with this pro- general population.	ons and cataract, and may lead to bil ith cracking, irritation and possible de r and may cause some concern follow duct is more likely to cause a sensitis	naness. rmatitis following. ring repeated or long-term occupational exposure. ation reaction in some persons compared to the	
	Prolonged or repeated skin contact may cause degreasi	ing, followed by drying, cracking and	skin inflammation.	
NV/ Tyro Gol	ΤΟΧΙCITY	IRRITATION		
NV Tyte Get	Not Available	Not Available		
	ΤΟΧΙCITY	IRRITATION		
polydimethylsiloxane	Dermal (rabbit) LD50: >3000 mg/kg ^[2]	Eye (rabbit): 100 r	ng/1h - mild	
	Oral (Rat) LD50; >35000 mg/kg ^[2]			
	ΤΟΧΙΟΙΤΥ	IRRITATION		
	dermal (rat) LD50: >2000 mg/kg ^[1]	Eye (rabbit): 0.75	mg/24h SEVERE	
	Oral (Rat) LD50; 1000 mg/kg ^[1]	Eye (rabbit): 100 r	ng	
lauryl alcohol, ethoxylated		Eye: adverse effe	ct observed (irritating) ^[1]	
		Skin (rabbit): 500	mg/24h mild	
		Skin (rabbit): 75 m	ng/24h mild	
		Skin: no adverse e	effect observed (not irritating) ^[1]	
	ΤΟΧΙΟΙΤΥ	IRRITATION		
water	Oral (Rat) LD50; >90000 mg/kg ^[2]	Not Available		
Legend:	1. Value obtained from Europe ECHA Registered Subst	ances - Acute toxicity 2.* Value obtair	ned from manufacturer's SDS. Unless otherwise	
	conjunctivitis. Asthma-like symptoms may continue for months or ever known as reactive airways dysfunction syndrome (RADS criteria for diagnosing RADS include the absence of pre	n years after exposure to the material S) which can occur after exposure to l vious airways disease in a non-atopic	ends. This may be due to a non-allergic condition nigh levels of highly irritating compound. Main individual, with sudden onset of persistent	
	asthma-like symptoms within minutes to hours of a documented exposure to the irritant. Other criteria for diagnosis of RADS include a reversibl airflow pattern on lung function tests, moderate to severe bronchial hyperreactivity on methacholine challenge testing, and the lack of minimal lymphocytic inflammation, without eosinophilia. Polyethers (such as ethoxylated surfactants and polyethylene glycols) are highly susceptible to being oxidized in the air. They then form complet mixtures of oxidation products. Animal testing reveals that whole the pure, non-oxidised surfactant is non-sensitizing, many of the oxidation products are sensitisers. The oxidization products also cause irritation. Humans have regular contact with alcohol ethoxylates through a variety of industrial and consumer products such as soaps, detergents and oth cleaning products. Exposure to these chemicals can occur through swallowing, inhalation, or contact with the skin or eyes. Studies of acute toxicity show that relatively high volumes would have to occur to produce any toxic response. No death due to poisoning with alcohol ethoxylate Both laboratory and animal testing has shown that there is no evidence for alcohol ethoxylates (AEs) causing genetic damage, mutations or cancer. No adverse reproductive or developmental effects were observed. Tri-ethylene glycol ethers undergo enzymatic oxidation to toxic alkoxy acids. They may irritate the skin and the eyes. At high oral doses, they m cause depressed reflexes, flaccid muscle tone, breathing difficulty and coma. Death may result in experimental animal. The material may produce severe irritation to the eye causing pronounced inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis.			
LAURYL ALCOHOL, ETHOXYLATED	mixtures of oxidation products. Animal testing reveals that whole the pure, non-oxidised oxidization products also cause irritation. Humans have regular contact with alcohol ethoxylates th cleaning products. Exposure to these chemicals can occ toxicity show that relatively high volumes would have to has ever been reported. Both laboratory and animal testing has shown that there cancer. No adverse reproductive or developmental effec Tri-ethylene glycol ethers undergo enzymatic oxidation t cause depressed reflexes, flaccid muscle tone, breathing. The material may produce severe irritation to the eye ca produce conjunctivitis. The material may cause skin irritation after prolonged or vesicles, scaling and thickening of the skin.	ylene glycols) are highly susceptible I surfactant is non-sensitizing, many of hrough a variety of industrial and com- cur through swallowing, inhalation, or occur to produce any toxic response. It is no evidence for alcohol ethoxylate ts were observed. It is no evidence for alcohol ethoxylate ts were observed. It is no evidence for alcohol ethoxylate ts were observed. It is a construct alkoxy acids. They may irritate g difficulty and coma. Death may resu- using pronounced inflammation. Rep	choline challenge testing, and the lack of minimal to being oxidized in the air. They then form complet of the oxidation products are sensitisers. The sumer products such as soaps, detergents and oth contact with the skin or eyes. Studies of acute No death due to poisoning with alcohol ethoxylate is (AEs) causing genetic damage, mutations or at the skin and the eyes. At high oral doses, they m ult in experimental animal. eated or prolonged exposure to irritants may a on contact skin redness, swelling, the production	
LAURYL ALCOHOL, ETHOXYLATED WATER	mixtures of oxidation products. Animal testing reveals that whole the pure, non-oxidised oxidization products also cause irritation. Humans have regular contact with alcohol ethoxylates th cleaning products. Exposure to these chemicals can occ toxicity show that relatively high volumes would have to has ever been reported. Both laboratory and animal testing has shown that there cancer. No adverse reproductive or developmental effect Tri-ethylene glycol ethers undergo enzymatic oxidation t cause depressed reflexes, flaccid muscle tone, breathing. The material may produce severe irritation to the eye ca produce conjunctivitis. The material may cause skin irritation after prolonged or vesicles, scaling and thickening of the skin. No significant acute toxicological data identified in literat	ylene glycols) are highly susceptible I surfactant is non-sensitizing, many o nrough a variety of industrial and cons cur through swallowing, inhalation, or occur to produce any toxic response. It is no evidence for alcohol ethoxylate the were observed. It is no evidence for alcohol ethoxylate the were observed. It is no evidence for alcohol ethoxylate the served. It is no evidence for alcohol ethoxylate the observed. It is no evidence for alcohol	choline challenge testing, and the lack of minimal to being oxidized in the air. They then form complet of the oxidation products are sensitisers. The sumer products such as soaps, detergents and oth contact with the skin or eyes. Studies of acute No death due to poisoning with alcohol ethoxylate as (AEs) causing genetic damage, mutations or the the skin and the eyes. At high oral doses, they m aft in experimental animal. eated or prolonged exposure to irritants may a on contact skin redness, swelling, the production	
LAURYL ALCOHOL, ETHOXYLATED WATER Acute Toxicity	mixtures of oxidation products. Animal testing reveals that whole the pure, non-oxidised oxidization products also cause irritation. Humans have regular contact with alcohol ethoxylates th cleaning products. Exposure to these chemicals can occ toxicity show that relatively high volumes would have to has ever been reported. Both laboratory and animal testing has shown that there cancer. No adverse reproductive or developmental effec Tri-ethylene glycol ethers undergo enzymatic oxidation t cause depressed reflexes, flaccid muscle tone, breathing The material may produce severe irritation to the eye ca produce conjunctivitis. The material may cause skin irritation after prolonged or vesicles, scaling and thickening of the skin. No significant acute toxicological data identified in literat	I surfactant is non-sensitizing, many of through a variety of industrial and com- cur through swallowing, inhalation, or occur to produce any toxic response. It is no evidence for alcohol ethoxylate ts were observed. To toxic alkoxy acids. They may irritate g difficulty and coma. Death may resu- using pronounced inflammation. Rep repeated exposure and may produce ture search. Carcinogenicity	to being oxidized in the air. They then form completed to being oxidized in the air. They then form completed to be oxidation products are sensitisers. The sumer products such as soaps, detergents and oth contact with the skin or eyes. Studies of acute No death due to poisoning with alcohol ethoxylate is (AEs) causing genetic damage, mutations or the the skin and the eyes. At high oral doses, they mult in experimental animal. eated or prolonged exposure to irritants may e on contact skin redness, swelling, the production	
LAURYL ALCOHOL, ETHOXYLATED WATER Acute Toxicity Skin Irritation/Corrosion	mixtures of oxidation products. Animal testing reveals that whole the pure, non-oxidised oxidization products also cause irritation. Humans have regular contact with alcohol ethoxylates th cleaning products. Exposure to these chemicals can oct toxicity show that relatively high volumes would have to has ever been reported. Both laboratory and animal testing has shown that there cancer. No adverse reproductive or developmental effect Tri-ethylene glycol ethers undergo enzymatic oxidation t cause depressed reflexes, flaccid muscle tone, breathin. The material may produce severe irritation to the eye ca produce conjunctivitis. The material may cause skin irritation after prolonged or vesicles, scaling and thickening of the skin. No significant acute toxicological data identified in literat	ylene glycols) are highly susceptible I surfactant is non-sensitizing, many of brough a variety of industrial and cons cur through swallowing, inhalation, or occur to produce any toxic response. It is no evidence for alcohol ethoxylate ts were observed. It to toxic alkoxy acids. They may irritate g difficulty and coma. Death may resulusing pronounced inflammation. Rep repeated exposure and may produce ture search. Carcinogenicity Reproductivity	to being oxidized in the air. They then form completed to being oxidized in the air. They then form completed to be a source of the oxidation products are sensitisers. The sumer products such as soaps, detergents and oth contact with the skin or eyes. Studies of acute No death due to poisoning with alcohol ethoxylate as (AEs) causing genetic damage, mutations or the the skin and the eyes. At high oral doses, they mult in experimental animal. eated or prolonged exposure to irritants may a on contact skin redness, swelling, the production	
LAURYL ALCOHOL, ETHOXYLATED WATER WATER Acute Toxicity Skin Irritation/Corrosion Serious Eye Damage/Irritation	mixtures of oxidation products. Animal testing reveals that whole the pure, non-oxidised oxidization products also cause irritation. Humans have regular contact with alcohol ethoxylates the cleaning products. Exposure to these chemicals can occ toxicity show that relatively high volumes would have to has ever been reported. Both laboratory and animal testing has shown that there cancer. No adverse reproductive or developmental effec Tri-ethylene glycol ethers undergo enzymatic oxidation the cause depressed reflexes, flaccid muscle tone, breathing. The material may produce severe irritation to the eye ca produce conjunctivitis. The material may cause skin irritation after prolonged or vesicles, scaling and thickening of the skin. No significant acute toxicological data identified in literat	I surfactant is non-sensitizing, many of hrough a variety of industrial and compound a variety of industrial and compound in the sensitizing of the sensitivity of the sensitizing of the sensitizing of the sensitizing of the sensitivity of th	tholine challenge testing, and the lack of minimal to being oxidized in the air. They then form complete to be being oxidized in the air. They then form complete to the oxidation products are sensitisers. The sumer products such as soaps, detergents and oth contact with the skin or eyes. Studies of acute No death due to poisoning with alcohol ethoxylate is (AEs) causing genetic damage, mutations or a the skin and the eyes. At high oral doses, they mult in experimental animal. eated or prolonged exposure to irritants may a on contact skin redness, swelling, the production	
LAURYL ALCOHOL, ETHOXYLATED WATER WATER Skin Irritation/Corrosion Serious Eye Damage/Irritation Respiratory or Skin sensitisation	mixtures of oxidation products. Animal testing reveals that whole the pure, non-oxidised oxidization products also cause irritation. Humans have regular contact with alcohol ethoxylates th cleaning products. Exposure to these chemicals can oct toxicity show that relatively high volumes would have to has ever been reported. Both laboratory and animal testing has shown that there cancer. No adverse reproductive or developmental effect Tri-ethylene glycol ethers undergo enzymatic oxidation t cause depressed reflexes, flaccid muscle tone, breathing. The material may produce severe irritation to the eye ca produce conjunctivitis. The material may cause skin irritation after prolonged or vesicles, scaling and thickening of the skin. No significant acute toxicological data identified in literat X	I surfactant is non-sensitizing, many of hrough a variety of industrial and cons cur through swallowing, inhalation, or occur to produce any toxic response. It is no evidence for alcohol ethoxylate ts were observed. It to toxic alkoxy acids. They may irritate g difficulty and coma. Death may resu using pronounced inflammation. Rep repeated exposure and may produce ture search. Carcinogenicity Reproductivity STOT - Single Exposure STOT - Repeated Exposure	tholine challenge testing, and the lack of minimal to being oxidized in the air. They then form complet of the oxidation products are sensitisers. The sumer products such as soaps, detergents and oth contact with the skin or eyes. Studies of acute No death due to poisoning with alcohol ethoxylate is (AEs) causing genetic damage, mutations or e the skin and the eyes. At high oral doses, they m att in experimental animal. eated or prolonged exposure to irritants may e on contact skin redness, swelling, the production	

SECTION 12 Ecological information

	Endpoint	Test Duration (hr)	Species	Value Sou	urce
NV Tyre Gel	Not Available	Not Available	Not Available	Not Not Available Ava	t ailable
polydimethylsiloxane	Endpoint	Test Duration (hr)	Species	Value Sou	urce
	Not Available	Not Available	Not Available	Not Not Available Ava	t ailable
lauryl alcohol, ethoxylated	Endpoint	Test Duration (hr)	Species	Value So	ource
	NOEC(ECx)	504h	Crustacea	0.144mg/l 2	
	EC50	48h	Crustacea	1.2mg/L 5	
	LC50	96h	Fish	1.5mg/l 2	
	Endpoint	Test Duration (hr)	Species	Value Sou	urce
water	Not Available	Not Available	Not Available	Not Not Available Ava	t ailable
Legend:	Extracted from Ecotox databas - Bioconcentrati	1. IUCLID Toxicity Data 2. Europe ECH e - Aquatic Toxicity Data 5. ECETOC Ac ion Data 8. Vendor Data	A Registered Substances - Ecotoxicological Ini uuatic Hazard Assessment Data 6. NITE (Japa	formation - Aquatic Toxicity 4. US EF an) - Bioconcentration Data 7. METI (PA, (Japan

Toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.

Wastes resulting from use of the product must be disposed of on site or at approved waste sites.

Toxic to bees.

For Siloxanes:

Environmental Fate: Siloxanes are used in cosmetics, wax, polishes, and to a minor extent in several other applications.

Atmospheric Fate: In the presence of nitrate ions, short chain siloxanes are broken down by sunlight to the level of silicate within days. The main source atmospheric siloxane release to the air is via evaporation.

Aquatic Fate: It is well accepted that polydimethylsiloxane fluids become permanent residents of sediment but should not have adverse environmental effects.

Microbial methylation plays important roles in the biogeochemical cycling of the metalloids and possibly in their detoxification. Many microorganisms (bacteria, fungi, and yeasts) and animals are now known to biomethylate arsenic, forming both volatile (e.g., methylarsines) and nonvolatile (e.g., methylarsonic acid and dimethylarsinic acid) compounds. Antimony and bismuth, also undergo biomethylation to some extent. Trimethylstibine formation by microorganisms is now well established, but this process apparently does not occur in animals.

For Surfactants: Kow cannot be easily determined due to hydrophilic/hydrophobic properties of the molecules in surfactants. BCF value: 1-350.

Aquatic Fate: Surfactants tend to accumulate at the interface of the air with water and are not extracted into one or the other liquid phases.

Terrestrial Fate: Anionic surfactants are not appreciably sorbed by inorganic solids.

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
lauryl alcohol, ethoxylated	LOW	LOW
water	LOW	LOW

Bioaccumulative potential

Ingredient	Bioaccumulation
lauryl alcohol, ethoxylated	LOW (LogKOW = 3.6722)

Mobility in soil

Ingredient	Mobility
lauryl alcohol, ethoxylated	LOW (KOC = 10)

SECTION 13 Disposal considerations

Waste treatment methods	
Product / Packaging disposal	Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked. A Hierarchy of Controls seems to be common - the user should investigate: Reduction Reuse Recycling Disposal (if all else fails) This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. DO NOT allow wash water from cleaning or process equipment to enter drains. It may be necessary to collect all wash water for treatment before disposal. In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first. Where in doubt contact the responsible authority. Recycle wherever possible. Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified. Dispose of by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or incineration in a licensed apparatus (after admixture with suitable combustible material).

SECTION 14 Transport information

Labels Required Marine Pollutant NO HAZCHEM Not Applicable

Land transport (ADG): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
polydimethylsiloxane	Not Available
lauryl alcohol, ethoxylated	Not Available
water	Not Available

Transport in bulk in accordance with the ICG Code

Product name	Ship Type
polydimethylsiloxane	Not Available
lauryl alcohol, ethoxylated	Not Available
water	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

polydimethylsiloxane is found on the following regulatory lists

Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 4 $\,$

lauryl alcohol, ethoxylated is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Industrial Chemicals Introduction Scheme Comparable Chemicals Table Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -Schedule 2

water is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

Australian Inventory of Industrial Chemicals (AIIC)

Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -Schedule 3

Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 4

Australian Inventory of Industrial Chemicals (AIIC)

National Inventory Status

National Inventory	Status	
Australia - AIIC / Australia Non-Industrial Use	Yes	
Canada - DSL	Yes	
Canada - NDSL	No (polydimethylsiloxane; lauryl alcohol, ethoxylated; water)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	No (polydimethylsiloxane)	
Japan - ENCS	No (polydimethylsiloxane)	
Korea - KECI	Yes	
New Zealand - NZIoC	Yes	
Philippines - PICCS	Yes	
USA - TSCA	Yes	
Taiwan - TCSI	Yes	
Mexico - INSQ	Yes	
Vietnam - NCI	Yes	
Russia - FBEPH	Yes	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.	

SECTION 16 Other information

Revision Date	12/10/2021
Initial Date	12/10/2021

SDS Version Summary

Version	Date of Update	Sections Updated
2.1	12/10/2021	Storage (suitable container)

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average PC-STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit. IDLH: Immediately Dangerous to Life or Health Concentrations ES: Exposure Standard OSF: Odour Safety Factor NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index AIIC: Australian Inventory of Industrial Chemicals DSL: Domestic Substances List NDSL: Non-Domestic Substances List IECSC: Inventory of Existing Chemical Substance in China EINECS: European INventory of Existing Commercial chemical Substances ELINCS: European List of Notified Chemical Substances NLP: No-Longer Polymers ENCS: Existing and New Chemical Substances Inventory KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals PICCS: Philippine Inventory of Chemicals and Chemical Substances TSCA: Toxic Substances Control Act TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas NCI: National Chemical Inventory FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

This document is copyright.

Apart from any fair dealing for the purposes of private study, research, review or criticism, as permitted under the Copyright Act, no part may be reproduced by any process without written permission from CHEMWATCH.

TEL (+61 3) 9572 4700.

